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Chemistry: Semester 2  
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PLAY. Terms in this set  
(...) Double

Replacement

Precipitation Reactions.

The positive ions in two  
compounds switch

places. A precipitation

reaction will produce

an insoluble salt as one

of the reactants. Mole.

Mole: A unit of

measurement used in

chemistry to express

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amounts of a chemical substance, defined as the amount of any substance that contains as atoms, molecules, ions, or electrons as there are atoms in 12 grams of pure C<sup>12</sup>, the isotope of ...

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Checklist As you

prepare for your final

exam use the following

list to monitor your

progress. Only check

off a learning goal

when you are confident

of your ability to

complete the task

described. Unit 0:

Being a Chemist

Recognize and apply

safe practices when

working in the lab.

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Honour Chemistry:

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(Semester 2) Part A:

Multiple Choice and

Numerical Response 1.

Which of the following

equations is associated

with the largest energy

change per mole of

fluorine? A.  $F_2(g) \rightarrow F_2$

(l) B.  $19\ 9\ F + 4\ 2\ He \rightarrow$

$1\ 0\ n + 22\ 11\ Na$  C.  $2$

$F_2(g) + 2\ H_2O(l) \rightarrow O_2$

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that you complete this review guide in its entirety Chemistry semester 2 final answers. You should also create your cheat sheet during this time, including all of the information and formulas. .

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liters, milliliters  $K = ^\circ C$

+ 273 g = grams Gas

constant  $R = 8314$

J/molK nm =

nanometers = 00821

Latm/molK atm =

atmospheres J, kJ =

joules, kilojoules

Standard T and P =

273K and 10 atm V =

volts mol = moles  $N_A$

=  $6.02 \times 10^{23}$  N

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2 Honors chemistry

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Which of the following

molecular properties is

a main component of

the chemical potential

energy of matter? A.

Vibrational motion C.

Movement from place

to place. B Honors

chemistry semester 2

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Intramolecular bonding

D. Rotation about the

molecules' centre of

mass.

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English; Published:

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