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negative charge. The charge for an  
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positive charge. The charge for a cation is written as a number followed by a plus sign. Chemical formula.

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Section 7. 1: Valence electrons electrons  
in the outer energy level o responsible  
for bonding with other atoms How do



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you know how many valence electrons  
an element has? o Look at the group  
number Group 1A = 1 valence electron  
Group 7A = 7 valence electrons Etc  
Electron Dot Diagrams a way of showing  
and keeping track of valence electrons o  
Written using the following steps: 1.

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**Answers**  
Chapter 7 Study Guide: Chemistry. To find the number of valence electrons in an atom of a representative element, simply look at its group number. A positively charged ion, or cation, is produced when an atom loses one or more valence electrons. Most ionic compounds are crystalline solids at room

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christymlove. BJU Chemistry Chapter 7  
Vocabulary. valence bond theory. sigma  
bond. pi ( $\pi$ ) bond. resonance. the  
concept that covalent bonds are formed  
when orbitals of di... a bond formed

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when orbitals overlap end to end on a bond axis. a bond formed when side-to-side orbital overlapping occurs.

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Chemistry Chapter 7 Study Guide. the force that holds two atoms together; may form by the attraction of a positive

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ion for a negative ion or by sharing electrons. the electrostatic force that holds oppositely charged particles together in an ionic compound.

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Chapter 7 Energy and Chemical  
Reactions 84 Study Guide for An

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Introduction to Chemistry Section Goals  
and Introductions Section 71 Energy  
Goals To introduce the terms energy,  
kinetic energy, and potential energy To  
introduce the Law of Conservation of  
Energy To describe the relationships

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Chlorofluorocarbons: A Chemical Success Story Gone Wrong CFCs and the Ozone Layer.

### **Chapter 7 Energy and Chemical Reactions**

CHAPTER 7 REVIEW. Chemical Formulas and Chemical Compounds. SECTION 1. SHORT ANSWER Answer the following



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questions in the space provided. 1. In a Stock system name such as iron(III) sulfate, the Roman numeral tells us. (a) how many atoms of Fe are in one formula unit.

## **7 Chemical Formulas and Chemical Compounds**

Chem 120 - Chapter 7 Study Guide.

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Chapter 7 - chemical reactions professor  
garima garg. University. Rio Hondo  
College. Course. Introduction To  
Chemistry (CHEM 120)

## **Chem 120 - Chapter 7 Study Guide - CHEM 120 - StuDocu**

Chapter 7 - Oxidation-Reduction  
Reactions<sup>87</sup> thus creating a voltaic cell,

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which is often called a battery. This section describes the fundamental components of voltaic cells and describes several different types.  
Chapter 7 Map

## **Chapter 7 Oxidation-Reduction Reactions**

Chapter 7 Energy and Chemical

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Reactions 84 Study Guide for An  
Introduction to Chemistry Section Goals  
and Introductions Section 71 Energy  
Goals To introduce the terms energy,  
kinetic energy, and potential energy To  
introduce the Law of Conservation of  
Energy To describe the relationships

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and the Royal Society of Chemistry, first introduced a Study Guide for Qualified Persons in 1978 based on Article 23 of 75/319/EEC. Further revisions were completed in 2000, 2004, 2006 and 2008. This revision was made effective from 1 February 2013.

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Model for Hydrogen o H has 1 electron  
and 1 proton o Electron orbits proton in  
different energy levels o Ground

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