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## Basic Stoichiometry Phet Lab Answer Key

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### Basic Stoichiometry Phet Lab Answer

LIV Brevis 11/2013 Nam Basic Stoichiometry Post-Lab Homework Exercise 1. Load the "Reactants, Products, and Leftover" simulation and work through each of the levels of the Game home, you can find the simulation by going to <http://phet.colorado.edu/ex/googling/phet>.

### Chies, Tif Name Sisted Basic Stoichiometry PET Lab ...

Basic Stoichiometry Phet Lab Essay Example | Graduateway. Basic Stoichiometry Phet Lab. Excess Reactant: A reactant that remains after a chemical reaction because there is nothing it can react with. Synthesis Reaction: Two or more chemical species combine to form a more complex product.

### Answers To Basic Stoichiometry Phet Lab

Name: Alexandria Jeremi Block 6 Basic Stoichiometry PhET Lab

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Let's make some sandwiches! Introduction: When we bake/cook something, we use a specific amount of each ingredient. Imagine if you made a batch of cookies and used way too many eggs, or not enough sugar. YUCK! In chemistry, reactions proceed with very specific recipes. The study of these recipes is stoichiometry.

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**\u200bAlexandria ...**

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## Phet Circuits Lab Worksheet Answers

Create your own sandwich and then see how many sandwiches you can make with different amounts of ingredients. Do the same with chemical reactions. See how many products you can make with different amounts of reactants. Play a game to test your understanding of reactants, products and leftovers. Can you get a perfect score on each level?

## Reactants, Products and Leftovers - PhET

Stoichiometry Lab. Stoichiometry February 28th, 2013 Abstract: The reactions of the Sodium Hydroxide and two acids, Hydrochloric Acid and Sulfuric Acid were performed. The heat given off by these two reactions was used to determine the stoichiometric ratio and the limiting reactants in each experiment.

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Basic Stoichiometry Phet Lab Answer Key Page 15/28 Read Online Basic Stoichiometry Phet Lab Answer Key(ANSWER 386.3g of  $\text{LiNO}_3$ ) 4) Using the following equation:  $\text{Fe}_2\text{O}_3 + 3 \text{H}_2 \rightarrow 2 \text{Fe} + 3 \text{H}_2\text{O}$ . Calculate how many grams of iron can be made from 16.5 grams of  $\text{Fe}_2\text{O}_3$  by the following equation.

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Reactants, Products and Leftovers

## **Reactants, Products and Leftovers**

Finally, students will be challenged to solve a basic stoichiometry problem! This product includes the PhET: Reactants, Products, and Leftovers Activity Guide and Answer Key in PDF format as well as a Getting Started Guide for teachers. The activity includes both color and grayscale files.

## **PhET Reactants, Products, and Leftovers Activity Guide ...**

Basic Stoichiometry Lab: Description Written as an introduction to mole-mole ratios and stoichiometry. Includes mole-mole post-lab homework. Does not included exercises covering mole-mass or mass-mass. This is my third draft of this lab, written 8/2010, editted 2/2011. If errors are found, please email me so that I may fix them.

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